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Basic Physics
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Formula - Definition, Equations, Examples

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2426 Electron = -1.602

$19 \times 10^{-19} \text{ C} = 9.11 \times$

10^{-31} kg Proton =

$1.602 \times 10^{-19} \text{ C} =$

$1.67 \times 10^{-27} \text{ kg}$

Neutron = $0 \text{ C} = 1.67$

$\times 10^{-27} \text{ kg}$ 6.022×10

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23 atoms in one atomic mass unit e is the elementary charge:
 $1.602 \times 10^{-19} \text{ C}$
Potential Energy,
velocity of electron: $PE = eV = \frac{1}{2}mv^2$
 $1 \text{ V} = 1 \text{ J/C}$
 $1 \text{ N/C} = 1 \text{ V/m}$
 $1 \text{ J} = 1 \text{ N}\cdot\text{m} = 1 \text{ C}\cdot\text{V}$

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